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Cool! I'am really happy

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so many fake sites. this is the first one which worked! Many thanks

REVIEW FOR CHEMISTRY FINAL

1. An example of a chemical change is
a. freezing water b. burning a match c. boiling CCl_4
d. dissolving alcohol in water e. stretching a rubber band
2. Give the laboratory instrument or instrument used to measure the following quantity:
mass _____
temperature _____
volume _____
density _____
length _____
3. A substance whose density is 4.80 g/cm³ occupies a volume of 12.3 cm³. What is the mass of the substance?
a. 6.00 g b. 6.00 g c. 60.0 g d. 6.00 g
4. From the following data determine the density of substance X.
Initial volume of water without substance X 10.0 mL
Mass of substance X 10.0 g
Final volume of water and substance X 17.0 mL
a. 1.4 g/mL b. 0.72 g/mL c. 0.72 g/mL d. 0.7 g/mL
5. Substance X has a specific heat of 0.10 J/g°C.
a. solid b. liquid c. gas
6. What is the correct answer when 480 g is divided by 0.0010 mL?
a. 4800 g/mL b. 48000 g/mL c. 480000 g/mL d. 4800 g/mL
7. How many oxygen atoms are there in $\text{Ba}(\text{NO}_3)_2$?
a. 7 b. 6 c. 5 d. 3
8. How much heat is required to raise the temperature of 25.0 g of water from 10.0°C to 40.0°C? The specific heat of water is 4.184 J/g°C.
a. 750 J b. 442 J c. 101 J d. 813 J

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